Chapter 1 Practice Questions

| 1) A good definition of chemistry is: | 1) | |
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| A) the science that seeks to understand what the universe does by studying interactions of molecules with atoms. | , | |
| B) the science that seeks to understand what living organisms do by studying the molecules that make up the organism. | | |
| C) the science that seeks to understand what matter does by studying what atoms and molecules do. | | |
| D) the science that seeks to understand the interactions of molecules for the sake of advancing human control over nature.E) none of the above | | |
| 2) Which statement about the scientific method is TRUE? | 2) | |
| A) The scientific method emphasizes observation and experimentation as the way to understand the world. | ۷) | |
| B) The scientific method emphasizes observation and reason as the way to understand the world. | | |
| C) The scientific method emphasizes scientific laws as the way to understand the world. D) The scientific method emphasizes reason as the way to understand the world. E) All of the above statements are false. | | |
| 3) Which statement about a hypothesis is TRUE? | 3) | |
| A) It is part of the scientific method. | | |
| B) It can be tested by experiments. | | |
| C) It is a tentative interpretation or explanation. | | |
| D) It has the potential to be proven wrong. | | |
| E) All of the above statements are true. | | |
| 4) The definition of a scientific law is: | 4) | |
| A) a way of learning that emphasizes observation and experimentation. | ŕ | |
| B) the underlying reason for a scientific theory. | | |
| C) the same as a hypothesis. | | |
| D) a number of similar observations generalized into a brief statement summarizing past | | |
| observations and predicting new ones. | | |
| E) none of the above | | |
| 5) What term best describes a brief statement that summarizes many past observations and predicts | 5) | |
| new ones? | ٠, | |
| A) experiment | | |
| B) scientific law | | |
| C) theory | | |
| D) hypothesis | | |
| E) none of the above | | |
| 6) Who discovered the law of conservation of mass? | 6) | |
| A) Antoine Lavoisier | ٠, | |
| B) John Dalton | | |
| C) John Dalton and Antoine Lavoisier | | |
| D) Nivaldo Tro | | |
| E) none of the above | | |

| 7) Which of the following is an example of a scientific law? | 7) |
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| A) When a can of soda pop is opened, a fizzing sound is heard. | |
| B) Flammable objects contain phlogiston. | |
| C) All matter is composed of small, indestructible particles called atoms. | |
| D) In a chemical reaction, matter is neither created nor destroyed. | |
| E) none of the above | |
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| 8) What is the definition of a scientific theory? | 8) |
| A) a brief statement that summarizes past observations and predicts future ones | · |
| B) a method of learning that emphasizes reason as the way to understand the world | |
| C) a model that explains the underlying reasons for observations and laws | |
| D) the equivalent of a scientific opinion which others may disagree with | |
| E) none of the above | |
| L) Hole of the above | |
| 9) Who formulated the atomic theory? | 9) |
| A) John Dalton and Antoine Lavoisier | ·) |
| B) Antoine Lavoisier | |
| C) Nivaldo Tro | |
| D) John Dalton | |
| E) none of the above | |
| =/ o- a a a | |
| 10) Which of the following is an example of an observation? | 10) |
| A) Reactions occur due to the transfer of electrons. | 10) |
| B) Flammable objects contain phlogiston. | |
| C) All matter is composed of small, indestructible particles called atoms. | |
| D) When a can of soda pop is opened, a fizzing sound is heard. | |
| E) none of the above | |
| | |
| 11) A sample of wood with a mass of 3.0 grams was burned in an open dish. The ashes weighed 0.2 | 11) |
| grams. What happened to the rest of the wood? | , |
| A) The remaining mass was destroyed. | |
| B) The remaining 2.8 grams was converted into gaseous compounds. | |
| C) The remaining 2.8 grams was converted into heat. | |
| D) Nothing, the burned wood simply has a mass of 1.2 grams. | |
| E) none of the above | |
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| 12) Which of the following is considered a hypothesis (as opposed to an observation)? | 12) |
| A) Fresh lava from a volcano is hot. | |
| B) Birds can fly because they have hollow bones. | |
| C) The Washington Monument is 555 feet tall. | |
| D) Spiders have eight legs. | |
| E) none of the above | |
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| 13) The key to success in chemistry is: | 13) |
| A) practice. | |
| B) mathematical skills. | |
| C) curiosity. | |
| D) commitment. | |
| E) all of the above | |

14) Which statement would NOT be an example of "quantification"?

- 14) _____
- A) The newest model of that car gets five more miles per gallon than last year's model.
- B) The water in the swimming pool feels colder than it was yesterday.
- C) The average high temperature this July was two degrees higher than last year's average high temperature.
- D) The price of milk went up ten cents per gallon.
- E) none of the above

Answer Key
Testname: PRACTICEQ_CH01

- 1) C 2) A 3) E 4) D 5) B 6) A 7) D 8) C 9) D 10) D 11) B 12) B 13) E

- 14) B